Preliminary communication

Synthesis of a hexasaccharide unit of a complex type of glycan chain of a glycoprotein*

TOMOYA OGAWA** and SATORU NAKABAYASHI***

The Institute of Physical and Chemical Research, Wako-shi, Saitama 351 (Japan) (Received March 19th, 1981; accepted for publication, April 20th, 1981)

In 1979, Baenziger and Fiete² reported that the complete structure of the glycan chain of fetuin, the major glycoprotein in fetal-calf serum, is 1. However, Svensson *et al.*³ proposed an isomeric structure concerning the branching mode for this glycan chain. Similar, complex structures have recently been reported for the glycan part of such glycoproteins as human α_1 -protease inhibitor⁴, membrane glycoprotein of vesicular-stomatitis virus⁵, α_1 -acid glycoprotein of human plasma⁶, and membrane glycoprotein of calf-thymocyte plasma⁷.

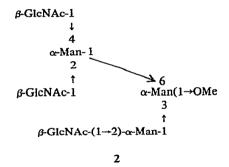
The presence in several glycoproteins of biological importance of unique, branching glycans having three such antennae as 1 has stimulated our efforts directed towards their chemical synthesis, and we now report a regio- and stereo-controlled synthesis of the hexasaccharide unit 2, a part of the structure of 1. Retrosynthesis of 2 revealed the necessary monosaccharide synthons 13, 8, 7, and 10 for the construction of 2 via the protected trisaccharide 12. As the preparation of synthons 13 (ref. 8), 8 (ref. 9), and 10 (ref. 10) has already been reported, an efficient route for the synthesis of glycosyl donor 7 should first be developed.

$$\alpha$$
-SA-(2 \rightarrow 3)- β -Gal-(1 \rightarrow 4)- β -GlcNAc-1
 4
 α -Man-1
 2
 \uparrow
 α -SA-(2 \rightarrow 6)- β -Gal-(1 \rightarrow 4)- β -GlcNAc-1
 β -Man-(1 \rightarrow 4)- β -GlcNAc-(1 \rightarrow 4)- β -GlcNAc-(

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^{**}To whom enquiries should be addressed.

^{***}Present address: Research Laboratories of Meiji Seika Kaisha, Ltd., Morooka-cho, Kohoku-ku, Yokohama, 222, Japan.



Allyl α -D-mannopyranoside 3 was regioselectively benzylated by the stannyl method^{9,11}, to give the 3,6-dibenzyl ether 4 in 54% yield[†]; $[\alpha]_D$ + 30.8° (c 0.39)^{††}; R_F 0.44 (1:1 toluene—EtOAc). Acetylation of 4 give diacetate 5 and deallylation of 5 in the presence of PdCl₂—NaOAc in 20:1 AcOH—H₂O for 14 h at 25° afforded hemiacetal 6 in 64% yield; $[\alpha]_D$ –18.9° (c 0.37); R_F 0.30 in 3:1 toluene—EtOAc; n.m.r. (CDCl₃): δ_H 5.31 (t, J 3 Hz, H-2), 5.14 (d, J 2 Hz, H-1), 5.10 (t, J 10 Hz, H-4), 3.90 (dd, J 3, 10 Hz, H-3), and 2.10 and 1.90 (2 s, 2 Ac). Transformation of 6 into the α -chloride 7 was achieved quantitatively by treatment of 6 with SOCl₂ in the presence of a trace of HCONMe₂ in Cl(CH₂)₂Cl for 14 h at 20°; 7: $[\alpha]_D$ +40.0° (c 0.27); R_F 0.52 in 3:1 toluene—EtOAc; δ_H (CDCl₃): 6.02 (d, J 2 Hz, H-1), 5.41 (dd, J 2, 4 Hz, H-2), 5.30 (t, J 10 Hz, H-4), 4.12 (dd, J 4, 10 Hz, H-3), and 2.10 and 1.89 (2 s, 2 Ac).

Glycosyl donor 7 having been synthesized, the route to the key intermediate, D-mannotrioside 12, became evident. Selective D-mannosylation at the primary hydroxyl group of 8 with 7 was achieved under the Hanessian-Banoub conditions¹², to afford a 50% yield of 9; $[\alpha]_D$ +29.7° (c 0.58); R_F 0.38 (3:1 toluene-EtOAc). Subsequent glycosylation of 9 with another D-mannosyl donor (10) afforded a 62% yield of the protected D-mannotrioside 11; $[\alpha]_D$ +35.2° (c 0.29); R_F 0.49 in 3:1 toluene-EtOAc. Zemplén deacetylation of 11 afforded triol 12; $[\alpha]_D$ +39.5° (c 0.39); R_F 0.30 in 1:1 toluene-EtOAc.

Glycosylation of 12 with donor 13 according to Lemieux et al. ¹³ gave a 35% yield of hexasaccharide 14, $[\alpha]_D$ +22.4° (c 0.38); R_F 0.70 in 25:2 CH₂Cl₂—acetone. Deacylation of 14 with BuNH₂ according to Durette et al. ¹⁴; and subsequent N-acetylation, led to the isolation of 15 in 20% yield; 15: $[\alpha]_D$ +19.4° (c 0.18, MeOH); R_F 0.21 in 80:24:1 CHCl₃—MeOH—H₂O. Hydrogenolysis of 15 in the presence of 10% Pd—C in aq. EtOH afforded the target hexasaccharide 2, $[\alpha]_D$ +14.7° (c 0.34, H₂O); R_F 0.20 in 10:30:3 CHCl₃—MeOH—H₂O. The ¹H-n.m.r. spectrum of 2 in D₂O at 80° showed signals for six anomeric protons: at δ 5.21 (d, 1 H, J 2 Hz), 5.00 (d, 1 H, J 2 Hz), and 4.80 (d, 1 H, J 1.5 Hz) for three anomeric protons for three α -D-mannopyranosyl

TYields are not optimized.

^{††}Unless otherwise noted, optical rotations were determined at 25° for solutions in CHCl₃.

residues, and 4.68 (d, 1 H, J 6 Hz), 4.66 (d, 1 H, J 5 Hz), and 4.63 (d, 1 H, J 5 Hz) for three anomeric protons for three 2-acetamido-2-deoxy- β -D-glucopyranosyl residues. The stereochemistry at the six anomeric carbon atoms was confirmed by 13 C-n.m.r. spectroscopy (D₂O at 80°), which showed three signals for anomeric carbon atoms of α -D-mannopyranosyl residues at δ 101.5 ($^{1}J_{CH}$ 172.0 Hz, C-1a), 100.0 ($^{1}J_{CH}$ 172.0 Hz, C-1c), and 97.5 ($^{1}J_{CH}$ 170.8 Hz, C-1b), and two signals for anomeric carbon atoms of 2-acetamido-2-deoxy- β -D-glucopyranosyl residues, at 101.9 ($^{1}J_{CH}$ 156.0 Hz, C-1d) and 100.4 ($^{1}J_{CH}$ 156.0 Hz, C-1e, 1f), in good agreement with the observation of Bock and Pedersen 15 for the value of $^{1}J_{CH}$ at anomeric carbon atoms.

In conclusion, a regio- and stereo-controlled synthesis of the hexasaccharide unit 2 could be achieved by employing properly protected mannotrioside 12 as a key intermediate, which, in turn, was prepared from regioselectively benzylated monosaccharide synthons 8 and 7. In close connection with our approach to the synthesis of 2, it may be noted that an approach towards complex types of glycans having two antennae was recently reported by Anarp and Lönngren¹⁶.

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REFERENCES

- 1 T. Ogawa and K. Sasajima, Carbohydr. Res., 93 (1981) 231-240.
- 2 J. U. Baenziger and D. Fiete, J. Biol. Chem., 254 (1979) 789-795.
- 3 B. Nilsson, N. E. Norden, and S. Svensson, J. Biol. Chem., 254 (1979) 4545-4553.
- 4 L. C. Hodges, R. Laine, and S. K. Chan, J. Biol. Chem., 254 (1979) 8208-8212.
- 5 C. L. Reading, E. E. Penhoet, and C. E. Ballou, J. Biol. Chem., 253 (1978) 5600-5612.
- 6 G. O. H. Schwarzmann, V. B. Hatcher, and R. W. Jeanloz, J. Biol. Chem., 253 (1978) 6983-6987.
- 7 H. Yoshima, S. Takasaki, and A. Kobata, J. Biochem. (Tokyo), 88 (1980) 819-827.
- 8 B. R. Baker, J. P. Joseph, R. E. Schaub, and J. H. Williams, J. Org. Chem., 19 (1954) 1786-1792; S. Akiya and T. Osawa, Yakugaku Zasshi., 77 (1957) 726-730.
- 9 T. Ogawa and M. Matsui, Carbohydr. Res., 62 (1978) C1-C4.
- T. Ogawa, K. Katano, and M. Matsui, Carbohydr. Res., 64 (1978) C3-C9; P. J. Garegg and L. Maron, Acta Chem. Scand. Ser. B, 33 (1979) 39-41.
- 11 T. Ogawa and M. Matsui, Carbohydr. Res., 51 (1976) C13-C18; Tetrahedron, (1981) in press.
- 12 S. Hanessian and J. Banoub, ACS Symp. Ser., 39 (1976) 36-63; Carbohydr. Res., 53 (1977) C13-C16.
- 13 R. U. Lemieux, T. Takeda, and B. Y. Chung, ACS Symp. Ser., 39 (1976) 90-115.

- 14 P. L. Durette, E. P. Meitzner, and T. Y. Shen, Tetrahedron Lett., (1979) 4013-4016; Carbohydr. Res., 77 (1979) C1-C4.
- K. Bock, I. Lundt, and C. Pedersen, Tetrahedron Lett., (1973) 1037-1040; K. Bock and
 C. Pedersen, J. Chem. Soc., Perkin Trans. 2, (1974) 293-297; Acta Chem. Scand., Ser. B,
 29 (1975) 258-264.
- 16 J. Arnarp and J. Lönngren, J. Chem. Soc. Chem. Commun., (1980) 1000-1002.